

Biodegradation Properties of Elektron 21 Magnesium Alloy Coated by Octacalcium Phosphate

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Biodegradation properties of magnesium and its alloys that can be used for implants are not satisfactory and cause serious problems. These problems can be solved by biodegradable surface coatings. Evaluation of biodegradation process of Mg-RE-Zr alloy surfaces after grinding and grinding followed by potential controlled electrodeposition of octacalcium phosphate (OCP) was investigated by electrochemical impedance spectroscopy measurements in this study. The potentiostatic electrodeposition treatment process was performed in water solution of 0.167 M CaCl_2 and 0.1 M $\text{NH}_4\text{H}_2\text{PO}_4$. The corrosion process on treated and nontreated samples after various exposure times was evaluated in 0.9% NaCl solution simulating body fluid environment at 37 °C. The significant increase of polarization resistance and time resistance against corrosion were found after electrochemical surface treatment.

Keywords: magnesium alloy, octacalcium phosphate, electrodeposition, corrosion resistance

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References

- [1] DeGARMO, P. E. (1979). *Materials and processes in manufacturing*, 5th ed. New York: Collin Macmillan, New York.
- [2] ZHANG, X., LI, Q., LI, L., ZHANG, P., WANG, Z., CHEN, F. (2012). Fabrication of hydroxyapatite/stearic acid composite coating and corrosion behavior of coated magnesium alloy. In: *Materials Letters*, pp. 76-78.
- [3] ZHANG, CH. Y., ZENG, R. CH., CHEN, R. S., LIU, CH. L., GAO, J. CH. (2010). Preparation of calcium phosphate coatings on Mg-1.0Ca alloy. In: *Transactions Nonferrous Metals Society of China*, pp. 655-659.
- [4] SÁNCHEZ-ENRÍQUEZ, J., REYES-GASGA, J. (2013). Obtaining $\text{Ca}(\text{H}_2\text{PO}_4)_2 \cdot \text{H}_2\text{O}$, monocalcium phosphate monohydrate, via monetite from brushite by using sonication. In: *Ultrasonics Sonochemistry*, pp. 948-954.
- [5] HARTWIG, A. (2001). Role of magnesium in genomic corrosion study. In: *Materials Science and Engineering C*, Vol. 33, pp. 675-679.
- [6] VOORT, G.F.V. (2004). *ASM Handbook - Metallography and Microstructures*. New York: ASM International, 7751184 pages.
- [7] SONG, Y.W., SHAN, D.Y., HAN, E.H. (2008). Electrodeposition of hydroxyapatite coating on AZ91D magnesium alloy for biomaterial application. In: *Materials Letters*, pp. 3276-3279.
- [8] HADZIMA, B., MHAEDE, M., PASTOREK, F. (2014). Electrochemical characteristics of calcium phosphatized AZ31 magnesium alloy in 0.9% NaCl solution. In: *Journal of Materials Science: Materials in Medicine*, Vol. 25, No. 5, pp. 1227-1237.
- [9] NOVÝ, F., JANEČEK, M., ŠKORÍK, V., MÜLLER, J., WAGNER, L. (2009). Very high cycle fatigue behaviour of as-extruded AZ31, AZ80, and ZK60 magnesium alloys. In: *International Journal of Materials Research*, Vol. 100, pp. 288-291.
- [10] FINTOVÁ, S., KUNZ, L. (2015). Fatigue properties of magnesium alloy AZ91 processed by severe plastic deformation. In: *Journal of the Mechanical Behavior of Biomedical Materials*, Vol. 42, pp. 219-228.
- [11] OMASTA, M., HADZIMA, B. (2015). Study of calcium phosphate (OCP) electrodeposition process on Elektron 21 magnesium alloy surface. In: *Materials Science Forum*, Vol. 818, pp. 115-120.
- [12] LORIMER, G., APPS, P., KARMIMZADECH, H., KING, J., (2003). Improving the performance of Mg-Rare Earth alloys by the use of Gd or Dy additions. In: *Materials Science Forum*, Vol. 419-422, pp. 279-284.